Antiferromagnetic and structural transitions in the superoxide KO₂ from first principles: A 2*p*-electron system with spin-orbital-lattice coupling

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KO₂ exhibits concomitant antiferromagnetic (AFM) and structural transitions, both of which originate from the open-shell 2p electrons of O₂⁻ molecules. The structural transition is accompanied by the coherent tilting of O₂⁻ molecular axes. The interplay among the spin-orbital-lattice degrees of freedom in KO₂ is investigated by employing the first-principles electronic structure theory and the kinetic-exchange interaction scheme. We have shown that the insulating nature of the high-symmetry phase of KO₂ at high temperature (*T*) arises from the combined effect of the spin-orbit coupling and the strong Coulomb correlation of O 2*p* electrons. In contrast, for the low-symmetry phase of KO₂ at low *T* with the tilted O₂⁻ molecular axes, the band gap and the orbital ordering are driven by the combined effects of the crystal field and the strong Coulomb correlation. We have verified that the emergence of the O 2*p* ferro-orbital ordering is essential to achieve the observed AFM structure for KO₂.

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Magnetism due to the correlated 2p electrons has attracted revived attention for the possibility of the new kinds of the magnetic informative materials.^{1,2} Some 2p magnetic oxides exhibit the structural phase transition concomitantly with the magnetic phase transition.³⁻⁵ Solid oxygen is a typical example, which has both antiferromagnetic (AFM) and structural phase transitions below 24 K.³ Alkali superoxides, AO₂ (A=Na,K,Rb), which are of our present interest, belong to another example.^{2,6} In AO₂, one alkali-metal atom provides one electron to an oxygen molecule, and thereby each O_2^- anion has nine electrons at the 2p molecular levels with the electronic configuration of $\sigma_g^2 \pi_u^4 \pi_g^3$ (see the inset of Fig. 2).^{1,7} The partially occupied antibonding π_g molecular states play the most important role in determining the electronic and magnetic properties of alkali superoxides.⁸ One hole in π_g generates the magnetic moment of $1\mu_B$ for each O_2^{-} . The degeneracy of the π_g level is expected to be lifted by lowering the crystal symmetry, as occurs due to the Jahn-Teller effect.⁸ In fact, it was suggested that, for KO_2 , the symmetry lowering would occur via coherent tilting of the O_2^{-} molecular axes, the so-called magnetogyration, which invokes the accompanying AFM ordering.4,5,9

At room temperature, KO₂ crystallizes in the tetragonal structure of CaC₂ type, in which the O₂⁻ molecular bond axes are parallel to the *z* axis [Fig. 1(a)].⁴ KO₂ retains this structure down to 197 K and exhibits the paramagnetic behavior. Upon cooling, O₂⁻ molecular bond axes seem to tilt uniformly by ~20° to have a lower crystal (monoclinic) symmetry. The magnetic phase is still paramagnetic down to 7 K. Below 7 K, the AFM ordering emerges in the triclinic crystal structure with the uniform tilting of O₂⁻ molecular bond axes by ~30°.⁴ According to neutron experiment,¹⁰ the AFM phase has the magnetic structure having opposite spin arrangements along the *z* direction between two oxygen layers of O1 and O2. This feature in KO₂ reflects the strong interplay among spin, orbital, and lattice degrees of freedom, as in rare-earth manganites.²

There have been several theoretical reports to study the coupled structural and magnetic transitions in KO_2 ,^{2,8,9,11,12}

in which the magnetic interactions between π_g orbitals were discussed with symmetry lowering. However, those studies were mostly qualitative and lacked the quantitative description of the electronic structures for the low-symmetry phase of KO₂. Even the direction of the tilted molecular bond axis is still uncertain between two possibilities. The first one is [R1] in Fig. 1(b), in which the molecular bonds are rotated around the [100] axis. The second one is [R2] in Fig. 1(c), in which the molecular bonds are rotated around [110] axis. Moreover, the band-structure study for the high-symmetry phase of KO_2 [Fig. 1(a)] in the local-density approximation (LDA) reveals that the degenerate π_g states do not split, resulting in the metallic nature, which is contradictory to the insulating nature of KO₂.² The spin-orbit (SO) effect was suggested to be important for splitting of π_g states in KO₂.² More recently, the Coulomb correlation effects of O 2p electrons were examined in similar 2p magnetic oxides, Rb_4O_6 and RbO₂.¹³⁻¹⁵ The magnetically frustrated insulating behavior of Rb₄O₆ and the orbital-ordered insulating state of RbO₂ were described by taking into account the Coulomb correlation of O 2p electrons.

In this Rapid Communication, we have investigated comprehensively the electronic structures of both high and low-



FIG. 1. (Color online) (a) The tetragonal [T] structure of KO₂ with unrotated O₂⁻ molecular bond axes. The molecular axes are along the *z* direction. K atoms are in blue (large spheres) and O atoms in red (small spheres). (b) [R1] with molecular bonds rotated by $\sim 30^{\circ}$ around the [100] (*x*') axis with bond axes along the *z*'-direction. (c) [R2] with molecular bonds rotated by $\sim 30^{\circ}$ around the [110] (*x''*) axis with bond axes along the *z*"-direction. There are two independent types of oxygen O1 and O2 for each structure.

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FIG. 2. (Color online) The DOS in the GGA for the FM phase of KO₂ with [T] structure. Dashed (blue) and solid (red) lines represent the total DOS and the partial DOS of oxygen 2p states, respectively. The positive and negative values correspond to the majority and minority-spin DOSs. Inset shows the 2p molecular levels of O₂⁻.

symmetry phases of KO₂ and explored the mechanism of the interplay between the spin and lattice degrees of freedom in correlated 2p electron systems. Further, we have examined the origins of the band-gap opening and the exchange interactions between O₂⁻ molecules in KO₂.

We have employed the full-potential augmented planewave band method¹⁶ implemented in WIEN2K package.¹⁷ For the exchange-correlation potential, the generalized gradient approximation (GGA) (Ref. 18) was used. We also incorporate the on-site Coulomb interaction U between the oxygen 2p electrons (GGA+U) (Ref. 19) and the SO effect as a second variational procedure (GGA+SO+U). For results below, U value was chosen to be 6.53 eV.²⁰

For [R1] and [R2] in Fig. 1, the O_2^- molecular bonds were assumed to be rotated uniformly by 30° from [T] structure around the [100] and [110] axes, respectively. The experimental lattice parameters a=4.030 Å and c=6.697 Å and the bond length $d_{oo}=1.306$ Å were used.²¹ The spin direction was chosen to be perpendicular to the molecular axis following the experiment.⁵

Figure 2 provides the density of states (DOS) in the GGA band calculation for the FM phase of KO₂ with the highsymmetry unrotated structure [T]. Oxygen 2*p* partial DOS (PDOS) shows clear molecular level splittings among σ_g, π_u, π_g and σ_u in agreement with literature.² The finite DOS at the Fermi level E_F in the minority-spin π_g states produces the half-metallic character for KO₂, which reflects the failure of the GGA. The degenerate π_g states, which are composed of antibonding π_x and π_y states, should split for an insulating phase of KO₂.

Figure 3 shows the DOS both in the GGA+U and GGA+SO+U for the FM phase of KO₂ with [T] structure. Note that the π_g states near E_F do not split in the GGA+U with U=6.53 eV. They become split only when the SO is included in the GGA+U: π_g split into $\pi_{m=1}$ and $\pi_{m=-1}$ which are mixed states of π_x and π_y . In the right of Fig. 3, the charge densities for the occupied $\pi_{m=1}$ states are plotted on the (001) and (010) planes. These charge densities manifest the azimuthally symmetric nature of $\pi_{m=1}$ states, which demonstrates that the splitting of π_g states originates from the SO effect. In fact, the large SO splitting results from the Cou-



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FIG. 3. (Color online) The PDOS of oxygen 2*p* states for the FM phase of KO₂ with [T] structure. The dashed (blue) and solid (red) lines represent the DOSs in the GGA+*U* and the GGA+SO+*U*, respectively (*U*=6.53 eV). The π_g states split into $\pi_{m=1}, \pi_{m=-1}$ in the GGA+SO+*U*. Charge densities of $\pi_{m=1}$ states are plotted on the *xy* (001) (the upper panel) and on the *xz* (010) plane (the lower panel).

lomb correlation, which localizes the 2p electrons to generate the substantial orbital magnetic moment in the π_g states. The magnitude of the orbital magnetic moment is as much as $0.570\mu_B$ per O₂⁻ molecule, which is comparable to that of the spin magnetic moment of $1.0\mu_B$. The direction of the orbital magnetic moment turns out to be almost parallel to the molecular axis, i.e., perpendicular to that of the spin magnetic moment. We have checked that the electronic structure of AFM KO₂ with [T] structure is close to that of FM KO₂, and so the band gap at E_F opens in the same way.

In the recent GGA+U calculation (without SO),¹⁵ the orbitally polarized insulating state was obtained for RbO₂ with [T] structure by initializing the orbital occupation matrix. The hybridization and electrostatic effects were proposed to lift the degeneracy between the different orbital-ordered states. We think, however, that the SO interaction, which has a sizable energy scale in superoxide (~20 meV),^{2,5} should be considered to obtain the correct ground state.

For KO₂ with [R1] and [R2] structures, the crystal-field effect will be activated due to tilting of O_2^- molecular axes toward K^+ . Figure 4(a) shows the local PDOSs and the spin densities in the GGA+SO+U for the AFM phase of KO_2 with [R1] structure. Notable feature in the DOS is the opening of band gap at E_F . The GGA band calculation does not produce the band gap, whereas both the GGA+U and the GGA+SO+U yield almost the same insulating band structure. This implies that, in producing the band gap, (i) the symmetry lowering just by tilting of molecular axes is not sufficient, (ii) the Coulomb correlation effect is essential, and (iii) the SO effect for [R1] is not as important as that for [T]. The orbital magnetic moment is almost quenched having only $0.002 \mu_B$ per O_2^- molecule. Therefore, it is the combined effect of the Coulomb U and the crystal field from K⁺ cations that splits the degenerate π_{e} states into the directional orbitals, $\pi_{x'}$ and $\pi_{y'}$ states. Interestingly, due to the complex interplay of the Coulomb U and the crystal-field effect in the molecular states, the splitting of π_{g} states occurs extraordinarily, i.e., the $\pi_{y'}$ - $\pi_{x'}$ order for the majority spin and the opposite $\pi_{x'}$ - $\pi_{y'}$ order for the minority-spin states. The halffilled molecular states $\pi_{y'}$ seem to be strongly affected by Coulomb U, while the fully occupied molecular states $\pi_{x'}$ seem to be weakly affected. The spin densities of unoccupied



FIG. 4. (Color online) (a) The local PDOS of the O 2*p* states in the GGA+SO+U (U=6.53 eV) for the AFM phase of KO₂ with [R1] structure. The spin densities of unoccupied $\pi_{y'}$ states are plotted on the (100) and (200) planes. Blue and red colors in the spin densities represent opposite spins. (b) The local PDOS of the O 2*p* states in the GGA+SO+U (U=6.53 eV) for the AFM phase of KO₂ with [R2] structure. The spin density of occupied minorityspin $\pi_{y''}$ states is plotted on the (110) plane. Black solid lines in the DOS plots represent the PDOS of K 4*p* states.

 π_g states plotted on the (100) and (200) planes demonstrate that they really correspond to the $\pi_{y'}$ states. The $\pi_{y'}$ states in each plane exhibit the FM spin and ferro-orbital (FO) orderings, while they exhibit the interplane AFM spin ordering. Namely, the AFM and FO orderings take place concurrently.

Similarly, Fig. 4(b) shows the local PDOSs and the spin densities in the GGA+SO+U for AFM KO₂ with [R2] structure. For [R2] too, the GGA does not produce the band gap, whereas both the GGA+U and the GGA+SO+U produce it. The SO effect is again negligible so that the orbital magnetic moment is quenched having only $0.008 \mu_B$ per O₂⁻ molecule. Degenerate π_g states are split into $\pi_{x''}$ and $\pi_{y''}$ due to the combined effect of the Coulomb U and the crystal field from K⁺. The spin density of occupied minority-spin π_g states plotted on the (110) plane demonstrates that they really correspond to the $\pi_{y''}$ states.

Now let us examine the magnetic interaction in KO₂ based on the above electronic structures. The implication of Fig. 4 is that, with the tilting of O_2^- molecular axes in the AFM [R1] and [R2] structures, the band-gap opening and the FO ordering in π_g states occur simultaneously. According to the spin-orbital model for 3d transition-metal (TM) oxides,²² the FO ordering of TM 3d states would induce the AFM superexchange interaction between two TM spins, and vice versa. In this respect, the results in Fig. 4 look reasonable. The superexchange, more generally, the kinetic exchange in KO₂ will take place through two channels: the direct hopping between two O₂⁻ molecules and the indirect hopping via K⁺ (see Fig. 5). The unoccupied K 4p states are located 4-6 eV above E_F (see Fig. 2), and so there is a possibility of hopping mechanism through O_2^{-} -K⁺- O_2^{-} . Indeed, as revealed in Fig. 4, there exists the non-negligible hybridization between the O 2p and K 4p states.

Table I presents the estimated exchange constants based on the microscopic calculation of the kinetic-exchange



FIG. 5. (Color online) (a) The in-plane exchange constants, J_1 and J_2 , on the *xy*-plane for KO₂ with [R2], where the unoccupied $\pi_{x''}$ and occupied $\pi_{y''}$ states are represented by red and green, respectively. (b) The interplane exchange constant J_3 on the (110) plane. The paths A and B correspond to the direct and indirect (via K⁺) hopping channels, respectively.

interaction²³ for KO₂ with the [R1] and [R2] structures. The 2p electrons in each O₂⁻ are assumed to have the atomicorbital states as in our band results for [R1] and [R2]. The kinetic-exchange interactions both from the direct and indirect hoppings were considered independently to calculate the exchange constant for each channel. For [R1], the exchange constants are obtained to be all AFM, and the dominant channel is the direct hopping between O₂⁻ anions along the y direction, which is AFM (J_2 =1.486 meV). The large J_2 results from the tilting of O₂⁻ molecular axis along the [010]. The interplane interactions from both direct and indirect channels yield smaller exchange constants (J_3 =0.209 and 0.027 meV). Then the resulting magnetic structure for [R1] will not be consistent with the experimental AFM structure which has alternating AFM spins along [001].

On the other hand, the kinetic-exchange interactions for [R2] turn out to be consistent with the experimental AFM structure. The dominant one is the interplane AFM interaction coming from the direct hopping (J_3 =0.680 meV). The in-plane exchange constants are 0.419 and -0.215 meV for direct and indirect channels, respectively. Hence the total exchange constants between two nearest neighbor (NN) O_2^- along x and y directions are 0.204 meV. For each O_2^- in [R2], there are eight interplane NN O_2^- and four in-plane NN O_2^- . As a consequence, J_3 , which is much larger than J_1 and J_2 will generate the interlayer AFM ordering along [001], as is consistent with the experimental AFM structure.

TABLE I. Magnetic exchange constants [meV] between O_2^- molecules for direct and indirect (via K⁺) hopping channels in the [R1] and [R2] structures of KO₂. J_1 and J_2 are the in-plane exchange constants along the *x* and the *y*-direction, and J_3 is the interplane exchange constant. Positive *J* represents the AFM interaction.

	J_1	J_2	J_3
R1 (direct)	0.353	1.486	0.209
R1 (via K ⁺)	0.023	0.023	0.027
R2 (direct)	0.419	0.419	0.680
R2 (via K ⁺)	-0.215	-0.215	0.005

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The in-plane FM interactions (-0.215 meV) coming from the indirect hopping via K⁺ play an important role in stabilizing the in-pane FM ordering. As shown in Fig. 5(a), out of two neighboring O_2^- molecules on the xy plane in [R2], the lobes of $\pi_{x''}$ orbital of one molecule are toward the intermediate K⁺, while those of the other molecule are away from K⁺ and thereby the lobes of $\pi_{y''}$ orbital are toward the intermediate K⁺. This arrangement of the orbitals suggests that the hopping takes place between the occupied $\pi_{y''}$ and unoccupied $\pi_{x''}$ orbitals of two neighboring O_2^- molecules. This is reminiscent of the resulting FM superexchange interaction between the fully occupied and half-filled orbitals in TM oxides in the framework of the Goodenough-Kanamori-Anderson (GKA) rule.²⁴ In contrast, for the interplane kinetic exchange in Fig. 5(b), the lobes of $\pi_{x''}$ orbitals of both O₂⁻ molecules are away from K⁺, and so the indirect kinetic-exchange results in the weak AFM interaction $(J_3 = 0.005 \text{ meV}).$

In conclusion, we have investigated electronic and

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magnetic structures of KO₂ superoxide, the strongly correlated 2p electron system. We have found that, for the correct description of the insulating electronic structure of KO₂, the SO coupling as well as the large Coulomb correlation is important for the high-symmetry phase, while, for the lowsymmetry phase, the crystal field from K⁺, as well as the large Coulomb correlation, is important. The concurrent AFM spin and FO orderings with the band-gap opening clearly demonstrate the strong coupling among the spinorbital-lattice degrees of freedom in KO₂. In the lowsymmetry phase of KO₂ with [R2] structure, the emergent FO ordering yields the kinetic-exchange interactions that are consistent with the experimental AFM structure.

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